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Full Length Research Paper

Flavonoid Compounds Composition and Antibacterial Activity of Origanum vulgare L. Sbsp. glandulosum (Desf) Ietswaart ethyl acetate extract

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ABSTRACT

The main objective of this work is the separation and identification of secondary metabolism of the subspecies *Origanum vulgare* L. Sbsp. *glandulosum* (Desf) Ietswaart, belonging to the Lamiaceae family. The use of the different chromatographic methods (column, paper, thin layer) permitted the isolation of a flavonoid: 5, 4'-dihydroxy-7-O- α -D-glucoside flavone (Apigenin7-O- α -D-glucoside). This compound is isolated for the first time from *Origanum vulgare* L. Sbsp. *glandulosum* (Desf) Ietswaart subspecies and also in *Origanum* kind. The structure of this compound was well established using state-of-the-art spectroscopic methods (UV, ¹H NMR, ¹³C NMR). Eventually, the antibacterial activity of ethyl acetate extract of the subspecies *Origanum vulgare* L. Sbsp. *glandulosum* (Desf) Ietswaart was tested with positive results, especially with Gram-negative bacteria *Pseudomonas aeruginosa* as well as with Gram-positive bacteria *Staphylococcus blanc*.

Key words: Origanum vulgare L. Sbsp. glandulosum (Desf) Ietswaart, Flavonoids, Ethyl Acetate

1. INTRODUCTION

Origanum is a genus of about 43 species of aromatic herbs in the family Lamiaceae [1,2], native from the Mediterranean Basin east of eastern Asia [1,3]. The genus includes some important culinary herbs, including *Origanum vulgare* L. Sbsp. *glandulosum* (Desf) Ietswaart.

Several species of this genus are used in traditional medicine in bitter stomachic. They are reported to have antiseptic, antispasmodic, as well as antioxidant effects [4-7]. Previous work on members of this genus revealed that the main constituents are flavonoids [8], essential oils [9], and terpenoids [10].

The main objective of this work is the separation and identification of secondary metabolism of the subspecies *Origanum vulgare* L. Sbsp. *glandulosum* (Desf) Ietswaart.

2. MATERIALS AND METHODS

2.1. Plant material

Aerial parts (flowers and leaves) of *Origanum vulgare* L. Sbsp. *glandulosum* (Desf) Ietswaart were collected in April 2005 from Jijel (texana), Algeria.



Fig. 1: Origanum vulgare L. Sbsp. glandulosum (Desf) Ietswaart

2.2. Extraction

The crucial first step for analysis of medicinal plants is to extract the desired chemical components from the plant materials for further separation and characterization.

Air-dried aerial parts of Origanum vulgare L. Sbsp. glandulosum (Desf) Ietswaart (1170g) were macerated three times with 70% MeOH and 30% water solution by replacing the solution every day with fresh solvent and one time with hot solvent. The hvdro-alcoholic solutions concentrated under were reduced pressure to dryness and the residue was dissolved in hot water being kept cold overnight. After filtration, the solution was successively aqueous extracted with dichloromethane, ethyl acetate, and n-butanol for two times for each solvent. Finally, the dichloromethane ethyl acetate and nbutanol extracts were concentrated to dryness. A brief summary of the extraction is described in Figure 2:

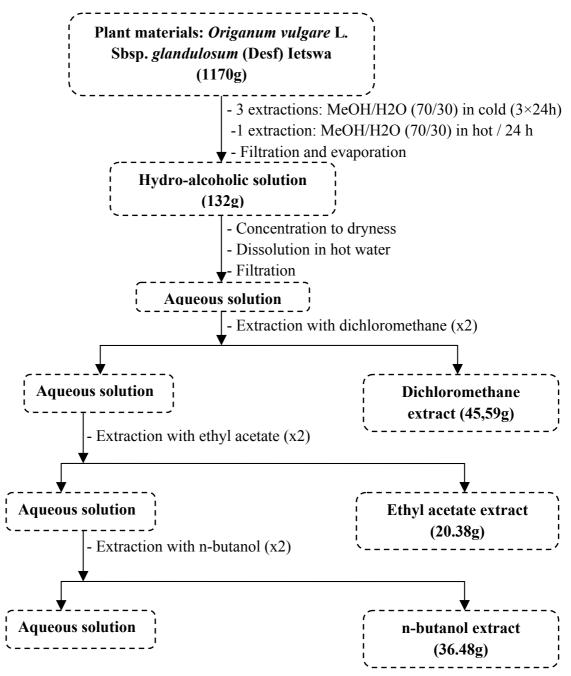


Fig. 2: Extraction Protocol of Origanum vulgare L. Sbsp. glandulosum (Desf) Ietswaart

2.3. Separation and purification of ethyl acetate extract compounds

The ethyl acetate extract (10.57g) was subjected to a SC₆ polyamide column chromatography being eluted with a gradient of Toluene/MeOH with increasing polarity. Twelve fractions were collected and compound A was obtained as a yellow precipitate from fraction 8 which was washed with methanol. This compound is identified for the first time in this species. The structure of compound *A* was determined by the usual spectroscopic methods (UV, ¹H NMR, ¹³C NMR).

2.4. Antibacterial activity

Moreover the antibacterial activities of the ethyl acetate extract of subspecies *Origanum vulgare* L. Sbsp. *Glandulosum* against four human pathogenic bacteria, including Gram-positive and Gramnegative bacteria, was carried out using disk diffusion method [11].

3. RESULTS AND DISCUSSION

3.1. Chromatographic comportment

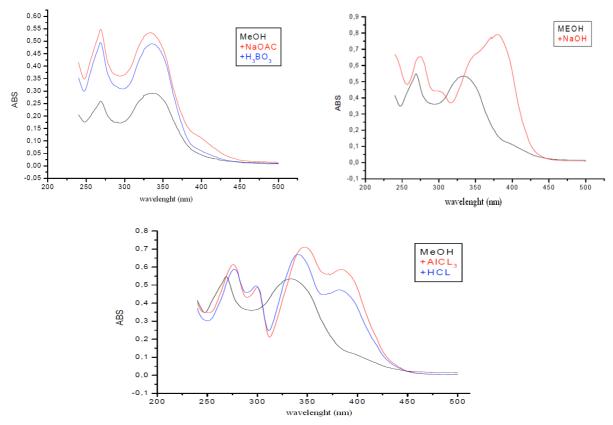
 Table 1: Chromatographic comportment of purified compound.

System	SI	SII
$\mathbf{R}_{\mathbf{f}}$	0.1	0.22
Spot color in ultraviolet	Without NH ₃	With NH ₃
light	Dark purple	Yellow

SI: (CH2Cl2/MeOH)(5/1)

SII: AcOH (15%)

The dark purple fluorescence and the R_f shifts show that the compound is a mono glycosides, either a flavone 3-H or a flavonol 3-OR substituted.



3.2. UV spectral data

Fig. 3: The UV spectra of compound A

The UV spectra of compound *A* obtained in this study were as follows: (MeOH, λ max, nm) gave bands at 335 and 269nm for band I and II, in addition of NaOH; 380,272 and 303nm, AlCl₃; 384, 275 and 299-347nm, AlCl₃/HCl; 381, 276 and 298-340nm, NaOAc; 333 and 269nm while NaOAc/H₃BO₃; 336 and 269nm. The UV spectrum in methanol and its changes after the addition of the customary shift reagents suggested that the compound is a flavone 3H with free hydroxyl groups at positions C4', C5, and a substitution in position 7 (7-OR) while shift reagents AlCl3, AlCl3/HCl exhibited band I absorption suggesting the absence of ortho-dihydroxyl group in A and B-ring [12,13]. The band-shifts in the UV spectra of compound A is tabulated in Table 2:

Chemical shift			Shift of observed
reagents	Band I (nm)	Band II (nm)	band (nm)
MeOH	335	269	-
NaOH	380	272	303
AlCl ₃	384	275	347-299
AlCl ₃ +HCl	381	276	340-298
NaOAc	333	269	-
NaOAc+H ₃ BO ₃	336	269	-

Table 2: The shift of bands in the UV spectra

3.3. NMR Spectroscopy

NMR is one the most powerful research techniques used to investigate the structure and properties of molecules. One of the main applications of NMR in flavonoid research is the structural elucidation of novel compounds about which nothing is known.

3.3.1. The 1H NMR spectrum

The ¹H NMR spectrum of the purified compound is illustrated in Figures 4, 5, and 6.

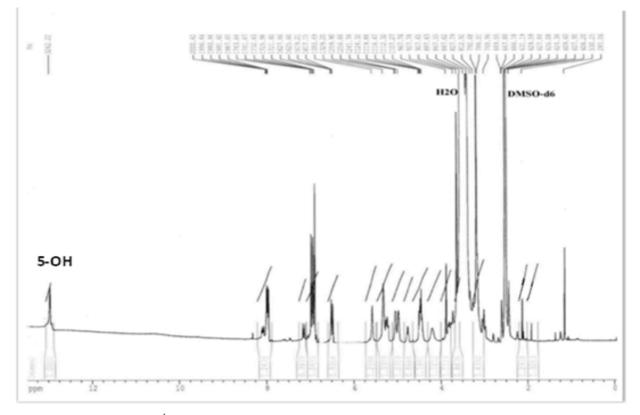


Fig. 4: ¹H NMR (DMSO-d₆; 250 MHz) spectrum of compound A

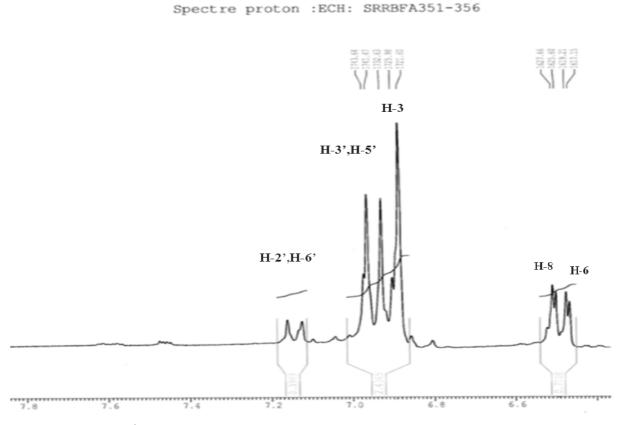


Fig. 5: ¹H NMR spectrum of compound *A* (enlarge the area 6.5- 8.00 ppm)

The spectral details of the purified compound are tabulated in Table 3.

Chemical moving	Integration	Multiplicity	Attribution
$\delta_{\rm H} (\rm ppm)$		J(Hz)	
6.48	1H	d (2.06)	H6
6.52	1H	d (2.06)	H8
6.9	1H	S	Н3
6.95	2H	d (8.84)	H3', H5'
7.15	2H	d (8.84)	H2', H6'
13	1H	S	5-OH

Table 3:	¹ H NMR	chemical	shifts
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The 1H NMR Spectra showed the presence of a singlet at $\delta = 13$ ppm, it was attributed to 5-OH

Two doublets at δ = 6.95ppm (2H, *J*=8.84 Hz) and δ = 7.15ppm (2H, *J*=8.84 Hz) was ascribed to H3', H5'and H2', H6'respectively, while the singlet at δ = 6.9ppm (1H) was attributed to H3.

Two doublets at δ = 6.48ppm (1H, *J*=2.06 Hz) and δ = 6.52ppm (1H, *J*=2.06 Hz) was ascribed to H6 and H8 respectively.

Sugar moiety at δ = 5.03ppm (1H, d, *J*=3.26 Hz, H1" glucose, α liaison to aglycon) [14].

Spectre proton :ECH: SRRBFA351-356

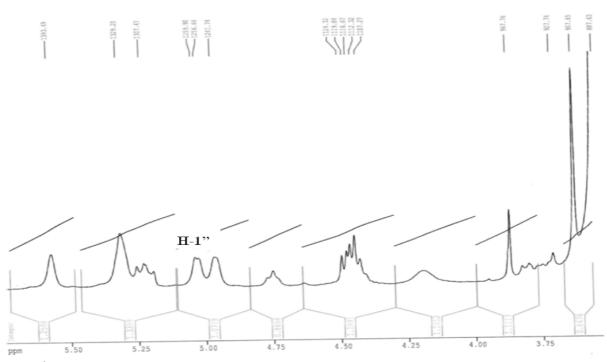
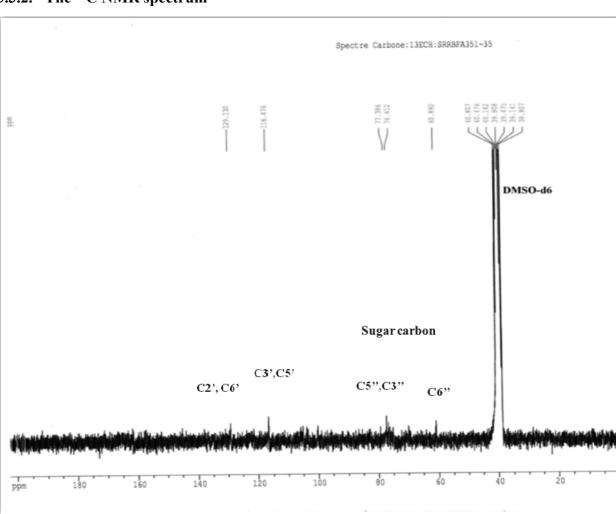


Fig. 6: ¹H NMR spectrum of compound *A* (Enlarge the area 3.75- 5.50 ppm)



3.3.2. The ¹³C NMR spectrum

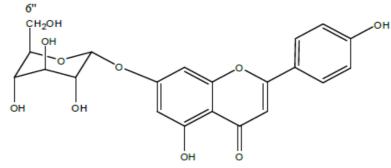
Fig. 7: ¹³C NMR (DMSO-d₆; 250 MHz) spectrum of compound A

The ¹³C NMR Spectra disclosed carbon signals, the results summarized in Table 3:

Carbon No.	¹³ C NMR chemical shifts (δ)	
C2', C6'	129.13	
C3', C5'	116.47	
C5", glucose	77.38	
C3", glucose	76.61	
C6'', glucose	60.89	

 Table 3: ¹³C NMR chemical shifts

The structure of this compound is:



compound 5,4'-dihydroxy-7-O- α -D-glucoside Fig. 8: Structural formulas of the flavone (Apigenin 7-O-α-D-glucoside)

3.4. Antibacterial activity

The bacterial strains were first grown on Muller Hinton medium (MHI) at 37°C for 24 hours prior seeding on the nutrient agar. A sterile 6mm diameter filter disk (Whatman paper # 3) was placed on the infusion agar seeded with bacteria and the extract suspended in water was dropped onto each paper disk (40 µL per disk) for all of the prepared concentrations (2mg/mL, 0.5mg/mL). The treated Petri disks were incubated at 37°C for 24h. Antibacterial activity was assessed by measuring the growthinhibition zone surrounding the disks. Each experiment was carried out in triplicate.

The results presented in Figure 9 show that the medium diameter of inhibition zone increases proportionally with the increase of flavonoids concentration. The obtained inhibition zone varied from 9.00 to 22.00mm.

The results were positive, especially with Gram negative bacteria Pseudomonas aeruginosa and Gram positive bacteria Staphylococcus blanc. However, no activity against Klebsielle pneumoniae at low concentration (0.5mg/ml) has been observed.

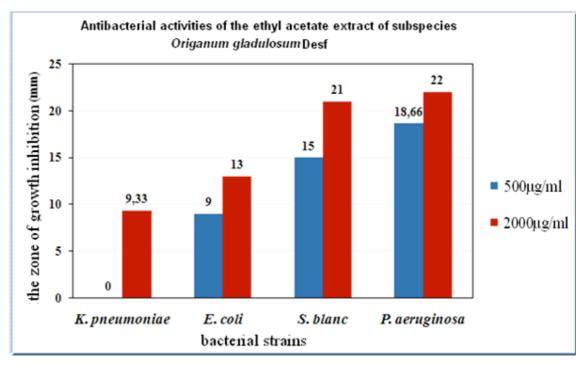


Fig. 9: Antibacterial activities of the ethyl acetate extract of subspecies *Origanum vulgare* L. Sbsp. *glandulosum*.

4. CONCLUSION

Antimicrobial activity of the ethyl acetate extract was reported. For the first time, one flavonoid has been isolated and identified from *Origanum vulgare* L. Sbsp. *glandulosum*.

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